

## Portulene Acetal, a Novel Minor Constituent of *Portulaca grandiflora* with Significance for the Biosynthesis of Portulal

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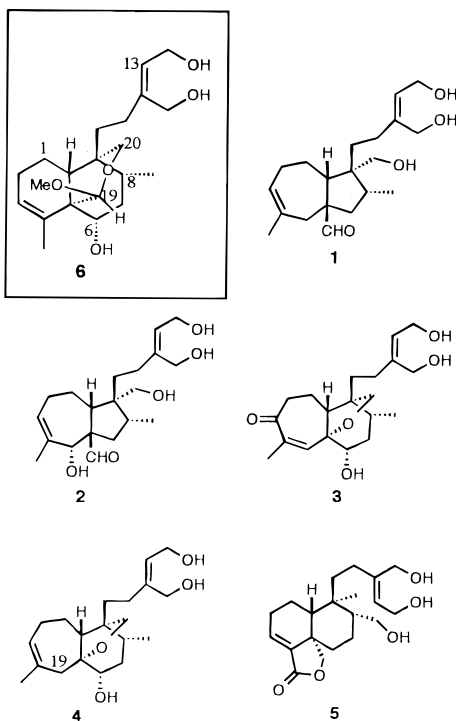
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Received May 12, 1997<sup>®</sup>

A new clerodane diterpenoid (**6**) with a caged hemiacetal ring has been isolated as a minor constituent from *Portulaca grandiflora* and its structure elucidated by extensive spectral studies. The compound might be of significance in the biosynthesis and in the chemosystematics of constituents of the genus *Portulaca*. The structural features of **6** provide important clues for the proposal of a chemically reasonable scheme for the biosynthesis of portulal (**1**).

Portulal (**1**) is a diterpene with a unique bicyclo[7.5]-decane ring system, isolated from *Portulaca grandiflora* Hook. (Portulacaceae) as a plant-growth regulator.<sup>1</sup> As a result of our interest in both its biosynthesis<sup>2</sup> and the chemosystematics of *Portulaca* species, we have carried out an extensive investigation on the constituents of *P. grandiflora* and related species.<sup>3</sup> We have characterized already six compounds<sup>4</sup> with a perhydroazulene skeleton and three compounds (e.g., **3**, **4**)<sup>5</sup> with a novel bicyclo[7.6]undecane ring system from *P. grandiflora*.



The isolation of portulide (**5**),<sup>6</sup> a clerodane diterpene, was reported by us, but subsequent cultivation studies

**Table 1.** NMR Data for Portulene Acetal (**6**) in CD<sub>3</sub>OD

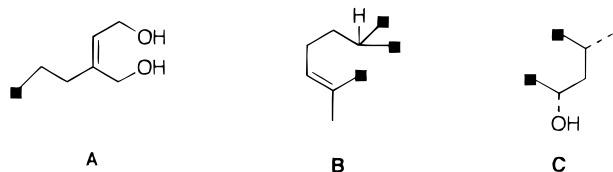
position	<sup>1</sup> H	<i>J</i> (Hz)	<sup>13</sup> C	HMBC with H no.
1a	1.53		20.28	
1b	1.94			
2	2.00		27.25	
3	5.50	br d (5.5)	126.50	1,18
4			138.77	18
5			47.03	1,3,7,18
6	3.53	dd (6.0,11.0)	77.20	7,19
7a	1.88		42.37	8,17
7b	2.02			
8	1.74	br quin. (6.7)	36.14	7,17,20
9			37.63	7,8,11,12,17
10	1.17		44.46	11,19,20
11a	1.17		30.21	12
11b	1.25	dt (5.0,14.0)		
12a	1.88		27.19	11,14,16
12b	2.01			
13			143.23	15,16
14	5.48	br t (6.7)	127.39	12,15,16
15	4.15	d (6.7)	58.71	
16	4.11	d (3.1)	60.14	14
17	0.96	d (6.7)	15.75	8,7
18	1.82	br s	22.71	3
19	4.88	s	100.36	6,20,21
20a	3.34	br d (11.6)	62.89	8,19
20b	3.48	br d (11.6)		
21	3.28	s	55.75	19

disclosed that this compound (renamed as portulide A) was originally a constituent of *Portulaca* cv. Jewel, and contamination of the commercial seeds of *P. grandiflora* with those of the latter variety caused an incorrect conclusion. In a continuation of studies on the minor constituents of *P. grandiflora*, we have confirmed the occurrence of a new clerodane diterpene, **6**, designated as portulene acetal, which might have significant implications in the biosynthesis of portulal (**1**).

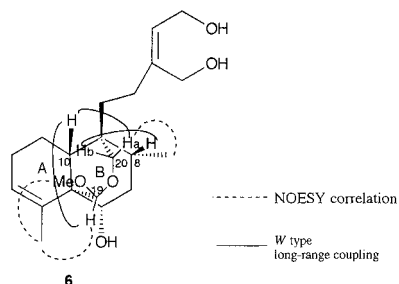
Portulene acetal (**6**) was obtained as colorless oil, [ $\alpha$ ]<sub>D</sub><sup>25</sup> +74.9° (*c* 0.25, MeOH), after repeated chromatography on Si gel columns and ODS columns of fractions obtained by successive partition with Et<sub>2</sub>O and EtOAc of the MeOH extract of fresh whole plants of *P. grandiflora*. Compound **6** was assigned the molecular formula C<sub>21</sub>H<sub>34</sub>O<sub>5</sub> on the basis of the [M]<sup>+</sup> peak in the HREIMS. In the IR spectrum, **6** exhibited hydroxyl absorption at 3400 cm<sup>-1</sup>. The signals in the <sup>1</sup>H- and

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<sup>®</sup> Abstract published in *Advance ACS Abstracts*, September 1, 1997.



**Figure 1.** Structural segments of compound **6** deduced on the basis of NMR experiments.



**Figure 2.** Significant NOESY correlations and *W*-type long-range couplings of **6** in the  $^1\text{H}$  NMR spectrum.

$^{13}\text{C}$ -NMR spectra ( $\delta$  3.28 and 55.75, respectively; see Table 1) indicated that an extra carbon atom on a diterpene skeleton could be attributed to a methoxy group. With the aid of  $^1\text{H}$ - $^1\text{H}$  COSY, DEPT, and  $^1\text{H}$ - $^{13}\text{C}$  COSY spectra, the presence of segments A-C was deduced (Figure 1). In addition, the presence of an oxymethylene group was revealed from the appearance of an AB quartet centered at  $\delta$  3.34 and 3.48 in the  $^1\text{H}$ -NMR spectrum and a signal at  $\delta$  62.89 (t) in the  $^{13}\text{C}$ -NMR spectrum. Both of the  $\text{H}_{\text{a}}\text{-20}$  and  $\text{H}_{\text{b}}\text{-20}$  signals were shown to have *W*-coupling with an angular proton, H-10, and a methine proton, H-8, respectively, by long-range  $^1\text{H}$ - $^1\text{H}$  COSY, HOHAHA, and homonuclear spin decoupling measurements being conducted. These structural features resembled closely those of portulene (**4**). The difference concerned ring A, which in **6** lacked an additional methylene group (C-19) adjacent to the trisubstituted double bond present in the case of **4**, as

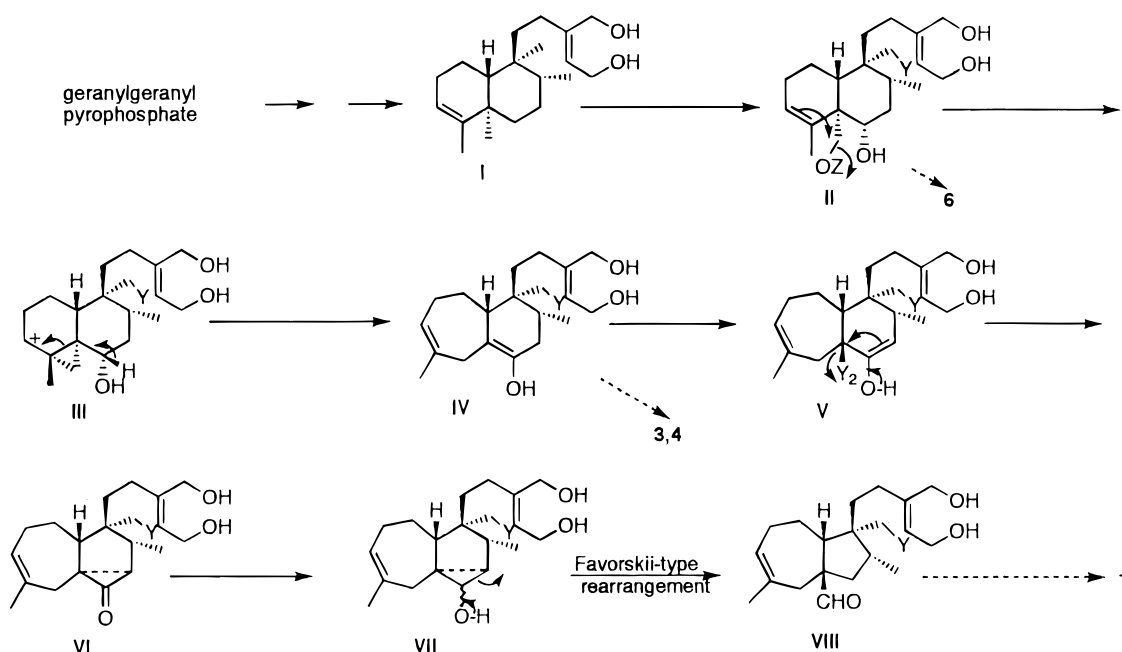
seen in segment B. Instead, the presence of an acetal methine group in **6** was indicated from the signals at  $\delta$  4.88 and 100.36 in the  $^1\text{H}$ -NMR and  $^{13}\text{C}$ -NMR spectra, respectively. These facts suggested that the C-19 carbon atom was part of an acetal ring structure with C-20, involving the methoxy group. HMBC correlations for **6**, which support all of the aforementioned deductions on the structure shown as **6** are given in Table 1. Thus, portulene acetal (**6**) was concluded to have a clerodane skeleton. The stereochemistry of the methoxy group was defined by 2D NMR measurements: *W*-coupling was observed between the acetal proton (H-19) and the angular proton (H-10) in the long-range  $^1\text{H}$ - $^1\text{H}$  COSY spectrum, and the H-19 and methoxy protons showed correlations with H-18 (vinyl methyl protons) in the NOESY spectrum, respectively, which indicated the relative configuration of **6** (Figure 2). Portulene acetal (**6**) might possibly be an artifact derived from the reaction of the corresponding hemiacetal, with the MeOH used in the extraction.

Thus, the co-occurrence in *P. grandiflora* of the clerodane diterpene **6** with a [6.6]-ring in addition to compounds with [7.6]- and [7.5]-ring systems has been confirmed. This finding supports the biosynthetic pathway, proposed by us, involving clerodane diterpenes in a linear path rather than in divergent one.<sup>3d</sup> Furthermore, the 6,19-oxygenation observed in **6** might be important in the mechanism of ring alteration in the biosynthetic process. With this consideration the detailed biosynthetic pathway shown in Scheme 1 is proposed. The intermediate II, a possible precursor of **6**, is transformed to a cyclopropylium cation III by homoallylic participation of C3-C4  $\pi$ -bond, which collapses to the [7.6]-bicyclo-intermediate IV. The ensuing Favorskii-type rearrangement would lead to the formation of portulal with the [7.5]-bicyclic ring system.

## Experimental Section

**General Experimental Procedures.** The following instruments were used: JASCO FT/IR-7000 (IR), JASCO DIP-370 polarimeter (optical rotation), JEOL JMS-

**Scheme 1.** Proposed Biosynthetic Pathway of Portulal (**1**)



HX-100 mass spectrometer (HRMS), and JEOL  $\alpha$ -500 NMR spectrometer ( $^1\text{H}$  and  $^{13}\text{C}$  NMR).

**Plant Material.** The material studied was cultivated from commercial seeds of *Portulaca grandiflora* at the botanical garden of the parks section of Osaka City. Whole plants, including leaves, stems and roots, were used.

**Extraction and Isolation.** Fresh plants (216 kg) were ground with MeOH and kept at room temperature for several weeks. After filtration, the filtrate was evaporated and partitioned successively with *n*-hexane, Et<sub>2</sub>O, and EtOAc. The combined Et<sub>2</sub>O and EtOAc extracts (170.2 g) were separated by Si gel column chromatography (EtOAc, EtOAc–EtOH, 97:3, 3:1, 1:1). The combined fraction (EtOAc–EtOH, 97:3 and 3:1, TLC guided, 45 g) was subjected to Si gel column chromatography (240–400 mesh, CHCl<sub>3</sub>–MeOH, 98:2) and MPLC on ODS (Fuji gel, C<sub>18</sub>, MeOH–H<sub>2</sub>O, 1:1) to give portulene acetal (**6**, 425 mg).

**Portulene Acetal (6):** colorless oil;  $[\alpha]_D^{25} +74.9^\circ$  (*c* 0.25, MeOH); IR (film)  $\nu_{\text{max}}$  3400 cm<sup>-1</sup>;  $^1\text{H}$  and  $^{13}\text{C}$  NMR, see Table 1; HREIMS (20 eV)  $[\text{M}]^+$  (obsd *m/z* 366.2412,

calcd 366.2406),  $[\text{M} - \text{OMe}]^+$  (obsd *m/z* 335.2198, calcd 335.2222),  $[\text{M} - \text{MeOH}]^+$  (obsd *m/z* 334.2140, calcd 334.2144).

## References and Notes

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NP970241T